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The Water Molecule: a Review. • Water is a

simple tri-atomic molecule, H₂O.

• Each O-H bond is highly polar, because of the high

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electronegativity of the oxygen (N, O, F, and Cl have high values)

- bond angle of water = 105° .

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Gravity. Created by.
Joseph_Wang6. Terms
in this set (233) Water
is an amazing liquid
that makes. plant and
animal life on Earth
Possible. In addition to

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water forming the
Earth's oceans water, it
also forms the.

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15.3 Heterogeneous
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15.3 Lesson Check -
Page 507: 21 Answer
Colloids cannot be
separated through
filtering because their
particles are smaller
than those of
suspensions and they
do not settle out over
time.

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Chemistry, Chapter 15, Water and Aqueous Systems. surface tension. surfactant. aqueous solution. solvent. the inward force or pull that tends

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to minimize the surface
ar... any substance
that interferes with
hydrogen bonding
between wa.... is water
that contains dissolved
substances.

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Chapter 15 "Water and
Aqueous Systems". use
these activities to
yourself study the
vocabulary and major

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concepts presented in this chapter. heterogeneous mixture containing particles that are small enough to remain dispersed in the solvent and do not separate on standing.

Quia - Chapter 15 "Water and Aqueous Systems"

Chapter 15 Water and Aqueous Systems 159
SECTION 15.1 WATER AND ITS PROPERTIES
(pages 445-449) This

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section describes the properties of water in the liquid and solid states and explains how hydrogen bonding affects the surface tension and vapor pressure of water.
Water in the Liquid State (pages 445–447)

SECTION 15.1 WATER AND ITS PROPERTIES (pages 445–449)

Water and Aqueous
Systems 445 ...

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insufficient fresh water.

Facts and Figures 446

Chapter 15
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with ChemASAP Liquid

water Hydrogen bond

Side view End view 2

Recall that water is a

simple triatomic

molecule, H_2O . The

oxygen atom forms a

covalent bond with

each of the hydrogen

atoms. Because of its

15.1 Water and Its Properties 15

1. Chapter 15 Chapter

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17 in your books "Water
and Aqueous Systems"

2. Section 15.1 Water
and its

Properties OBJECTIVES:

- Explain the high surface tension and low vapor pressure of water in terms of the structure of the water molecule and hydrogen bonding.

Chemistry - Chp 15 - Water and Aqueous Systems - Notes

Title: Chapter 15

Page 16/26

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Review Water and Aqueous Systems 1
Chapter 15 Review
Water and Aqueous
Systems. Pre-AP
Chemistry ; Charles
Page High School ;
Stephen L. Cotton; 2
Chapter 15 Review.
Surface tension is the
____. How does the
surface tension of
water compare with
the surface tensions of
most other liquids?
Which type of
mixture(s) exhibit the

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**PPT - Chapter 15
Review Water and
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Chapter 15 - Water and
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15.2 Homogeneous

Aqueous Systems -

15.2 Lesson Check -

Page 501: 11 Answer

All ionic compounds
are electrolytes

because they

dissociate into ions and
conduct an electrical

current in aqueous

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solution or in the
molten state.

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together with Useful
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want to offer
everything you need in
one authentic in
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Chapter 15 - Water and
Aqueous Systems -
15.1 Water and Its
Properties - 15.1
Lesson Check - Page
493: 2 Answer The
structure of ice is a
regular open
framework of water

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molecules arranged
like a honeycomb.

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Chapter 15 "Water and
Aqueous Systems"
Tools. Copy this to my
account; E-mail to a
friend; Find other
activities ... a relatively
strong intermolecular
force responsible for
water's high surface
tension: suspension: a

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mixture from which
particles settle out
upon standing:
aqueous solution: a
solution in which the
solvent is water:
solvent: the ...

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Page 501: 12 Answer

The forces holding the water molecules in hydrates are not very strong, so the water is easily lost and regained

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SECTION 15.1 WATER
AND ITS PROPERTIES

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(pages 445-449)

CHAPTER 15, Water and
Aqueous
Systems(continued) 6

Practice Problem
Circle the letter next to
each sentence that
describes a result of
the surface tension of
water a In a full glass
of water, the water
surface seems to bulge
over the rim of the
glass b Water beads up
into

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